# **MSDS** Document

## **Product WEEKLY BOOST**

## 1. Chemical Product and Company Identification

## **Product WEEKLY BOOST**

## MSDS ID MSDS2290

Manufacturer

Haviland Consumer Products, Inc. 421 Ann Street N.W. Grand Rapids, MI 49504 Phone Number
(616) 361-6691
Emergency Phone
CHEMTREC (800) 424-9300
CHEMTREC International (703) 527-3887
Revision Date 10/8/2014



## 2. Composition and Information on Ingredients

Ingredient Proprietary	CAS Number Proprietary	Weight %	ACGIH TLV	PEL	STEL
"Oxone" Monopersulfate Compound	70693-62-8		Not Established	Not Established	

## 3. Hazard Identification

## **Emergency Overview**

Material is corrosive to skin and eyes, and a nose and throat irritant. May cause allergic skin reactions in sensitive individuals. Ingestion may cause inflammation and damage to the lining of the stomach, resulting in bleeding.

#### **Health Effects**

Skin contact with moisture or perspiration may cause skin burns or ulceration; temporary body hair loss may occur in contacted areas. Eye contact may cause eye corrosion or ulceration. Inhalation may cause nose bleeds and irritation of the upper respiratory passages with coughing and discomfort. Ingestion may cause gastritis possibly progressing to necrosis or hemorrhage.

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## Carcinogenicity

None of the components present in this material are listed by IARC or NTP as a carcinogen.

## **Reproductive Effects**

None known

#### **Routes of Entry**

Inhalation, Ingestion, Skin or Eye Contact

## 4. First Aid Information

#### Inhalation

Remove to fresh air immediately. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Call a physician.

#### **Skin Contact**

Flush with plenty of water for at least 15 minutes while removing contaminated clothing. Get medical attention.

## **Eye Contact**

Flush immediately with plenty of water for at least 15 minutes, get medical attention.

## Ingestion

Call a physician immediately. Do not induce vomiting. Dilute by drinking water or milk. Never give anything by mouth to an unconscious person.

## 5. Fire Fighting Measures

Flash Point N/A

## **Extinguishing Media**

Use water only. Do not use dry chemicals, carbon dioxide, or foam.

### **Special Fire Fighting Procedures**

Will release oxygen when heated, intensifying a fire. Acidic mist may be present; self contained breathing apparatus should be used.

## **Unusual Fire Hazard**

Grinding or intensive mixing may cause decomposition with liberation of heat and oxygen; ignition of oxidizable material if present may occur.

### 6. Accidental Release Measures

#### Spill and Leak Procedures

Use appropriate personal protective equipment during clean-up. Sweep up solid. Flush liquid spills with low pressure water.

## 7. Handling and Storage

#### **Handling Procedures**

Do not inhale. Do not get in eyes, on skin or clothing. Wash thoroughly after handling. Wash

clothing after use.

## **Storage Procedures**

Store indoors in a cool dry area. Keep containers closed when not in use. Ensure thorough ventilation of work areas. Avoid all sources of ignition - heat, sparks, and open flame.

## 8. Exposure Controls and Personal Protection

## **Engineering Controls**

Provide ventilation sufficient to maintain exposure below the recommended limits.

#### **Eye Protection**

Safety glasses or goggles required.

#### **Protective Gloves**

Rubber or other impervious materials.

#### **Respiratory Protection**

Use NIOSH approved respiratory protection where airborne level exceeds appropriate occupational exposure limit.

#### **Other Protective Equipment**

A safety shower and eye bath should be available. Use an impervious body covering and boots.

## 9. Physical and Chemical Properties

Physical State Solid Specific Gravity 1.3

Color/Appearance White granules or powder

Odor Odorless
Boiling/Cond. Point Decomposes
Melting/Freezing Point Decomposes
Solubility 21% @ 20°C (68°F)

Vapor Density Not volatile

Vapor Pressure Nil

## 10. Stability and Reactivity

### **Chemical Stability**

Stable

#### **Conditions to Avoid**

Excessive heat.

### **Incompatible Materials**

Mixing with compounds containing halides or active halogens (bromine, chlorine, iodine) can cause the release of the respective halogen gas, if moisture is present. Mixing with dry or concentrated chlorine-containing chemicals, such as hypochlorites, dichlor, trichlor, or salt, may cause the release of chlorine gas. Mixing with cyanides can cause release of hydrogen cyanide gas. Mixing with heavy metal salts can cause decomposition with release of oxygen and heat.

## **Hazardous Decomposition Products**

Decomposes when heated or dampened, releasing oxygen and heat of decomposition.

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### **Hazardous Polymerization**

Will not occur.

## 11. Toxicological Information

No information available.

## 12. Ecological Information

No information available.

## 13. Disposal Considerations

## **Waste Disposal**

Comply with federal, state and local regulations. Dispose of waste in an approved chemical landfill. Large quantities should be neutralized with soda ash.

## 14. Transportation Information

### **DOT Shipping Name**

Corrosive solid, acidic, inorganic, n.o.s. (Monopersulfate compound)

#### **Hazard Class**

8, Corrosive

## **UN Number**

UN3260

## **Packing Group**

PG II

NOTE: Corrosive materials in Packing Group II, in inner packagings not over 1.0 L (0.3 gallon) for liquids or 1.0 kg (2.2 pounds) for solids, packed in strong outer packagings, may be renamed "Consumer Commodity" and reclassed as ORM-D material. Total package may not exceed 30 kg (66 pounds) gross weight.

## 15. Regulatory Information

#### **TSCA**

Hazardous Component(s) subject to reporting on the TSCA List.

#### **WHMIS**

Hazardous Component(s) subject to WHMIS Ingredient Disclosure.

## 16. Other Information

This information is based upon technical information believed to be reliable. It is subject to

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revision as additional knowledge and experience is gained.

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